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Title: Refrigeration And Air Conditioning Stoecker Solution Manual Author: wiki.ctsnet.org-Stephan

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3-1 Calculate the specific volume of an air-vapor mixture in cubic meters per kilogram of dry air when the following conditions prevail: $t = 30\text{ C}$, $W = 0.015\text{ kg/kg}$, and $p_t = 90\text{ kPa}$. Solution: Equation 3-4. $t_s = t_a = t_p = t_R = T = 30\text{ C} + 273 = 303$

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$R = 287 \text{ J/kg}\cdot\text{K}$ $P = 90 \text{ kPa} = 90,000 \text{ Pa}$
Equation 3-2 $t_2 = 30^\circ\text{C}$
 $t_1 = -10^\circ\text{C}$
 $W = 1.35 - 0.15 \text{ p s}$

or **CLICK HERE**

Solution: $q = mc_p(t_2 - t_1)$ $m = 2.5 \text{ kg/s}$
 $c_p = 1.0 \text{ kJ/kg}\cdot\text{K}$ $t_2 = 30^\circ\text{C}$
 $t_1 = -10^\circ\text{C}$ CHAPTER 2 -
THERMAL PRINCIPLES Page 2 of 5
Then, $q = (2.5)(1.0)(30 + 10) \text{ kJ/s}$
 $= 100 \text{ kW}$ - - - Ans. 2-3.

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