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a chemical change in which one element replaces a second element in a compound; displacement reaction Activity Series a list of elements in order of decreasing activity

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Stoker, Introduction to Chemical Principles | Pearson
• A skeleton equation is a chemical equation that does not indicate the relative amounts of the reactants and products. Like a recipe with no amounts listed.

Chapter 11
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11.2 Types of Chemical Reactions- 13 A decomposition reaction is a chemical change in which a single compound breaks down into two or more simpler products. • Decomposition reactions involve only one reactant and two or more products. • The products can be any combination of elements and compounds. • Most decomposition reactions require energy