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~~\_\_\_ Ca(PO)(s) \_\_\_ HSO(aq) \_\_\_ CaSO(s) \_\_\_ HPO(a<sub>3</sub> 4 2 + 2 4 4 + 3 4 q) Coefficient of H<sub>2</sub>SO<sub>4</sub> (aq) Coefficient of H<sub>3</sub>PO<sub>4</sub> (aq) A. 1 2 B. 2 3 C. 3 1 D. 3 2 3. 7.102 g of 1 NaSO (142.04gmol)<sup>2</sup> 4 M = - is dissolved in water to prepare 0.5000 dm<sup>3</sup> of solution. What is the concentration of Na<sub>2</sub>SO<sub>4</sub> in mol dm<sup>-3</sup>? A. 2.500 × 10<sup>-2</sup> B. 1.000 × 10<sup>-1</sup> C. 1.000 ...~~

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— Chemistry in industry and technology 10 – 13

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HI CHEMISTRY NOTES. TOPIC 1: QUANTITATIVE CHEMISTRY. TOPIC 2: ATOMIC STRUCTURE ... 9: REDOX PROCESSES. TOPIC 10: ORGANIC CHEMISTRY. TOPIC 11: MEASUREMENT AND DATA PROCESSING. OPTION B: BIOCHEMISTRY. THE HELPFUL CHEMISTRY RESOURCES I FOUND USEFUL. RadioChemistry: IB Online Teacher; ... past papers. specimen papers. Powered by Create your own ...

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