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~~Oklahoma State~~

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instructions on how to use this software, and
examples of guided inquiry and open-ended
inquiry activities click here.. We are
interested in your assessment of the

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activities you did with this software.

~~"Gas Law Program" — Oklahoma State
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Gas Law Simulation This guided inquiry experiment is organized along the same structure as the macroscopic experience described above. In this activity the students begin with a 2 page description of what the screen looks like and an invitation to play with/explore the different parts of the screen.

~~MoLE — Oklahoma State University Stillwater
Simulation . K. Haines (Oklahoma State
University), M.R. Abraham (University of
Oklahoma) and J.I. Gelder (Oklahoma State
University) ... open a page with a link to
the Pressure Temperature Relationship Guided
Inquiry activity and to two mp4 movies of the
Gas Law simulation (Java version) that can be
used to answer the questions in the guided
...~~

~~MoLE Web Inquiry Activities and Simulations~~

2. Use the Kinetic Molecular Theory to explain the gas laws. Describer ideal gas behavior in terms of the Kinetic Molecular Theory. 3. Apply the relationship of P , V , T , and n and Dalton's Law to mixtures of gases. 4. Determine the relationship among molar mass, molecular speeds (Graham's law), average kinetic energy, and temperature of gases.

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"Gas Law Program" - Oklahoma State University-Stillwater A. Open the Gas Law Simulation program and observe and describe, in the space below, the activity in the Gas Sample window. Consider using some or all of the following terms in your description: particles, atoms, molecules, collisions, velocity, energy, force.

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Gas Law Simulator. Click the image below and you can play with our gas simulator. You can control the relative amounts of two gases, A and B with the pump handle. You can release gas with the release valve. The temperature can be raised and lowered, plus you can have volume locked or change.

~~Gas Law Simulator University of Texas at Austin~~

Pump gas molecules to a box and see what happens as you change the volume, add or remove heat, and more. Measure the temperature and pressure, and discover how the properties of the gas vary in relation to each other. Examine kinetic energy and speed histograms for light and heavy particles. Explore diffusion and determine how concentration, temperature, mass, and radius affect the rate of ...

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