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Chapter 6 Answers Page 195 #1. Use the tables on page 420 (heat of dissolution) and 422 (polyatomic ions) to help you answer these questions. a) $\text{HClO}_3 (\text{s}) \rightarrow \text{H}^+ (\text{aq}) + \text{ClO}_3^- (\text{aq})$?H= b) $\text{HI} (\text{s}) \rightarrow \text{H}^+ (\text{aq}) + \text{I}^- (\text{aq})$?H= c) $\text{KNO}_3 (\text{s}) \rightarrow \text{K}^+ (\text{aq}) + \text{NO}_3^- (\text{aq})$... <https://studylib.net/doc/5610530/chemistry-chapter-6-answer-key>.

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Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. _____ In metals, the valence electrons are considered to be (a) attached to particular positive ions.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 1
SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

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~~/ Overview~~

SECTION 6.1 Nature favors arrangements in which potential energy is minimized. For example, a boulder is less likely to balance at the top of a hill than it is to roll to the bottom of a valley. The boulder at the top of the hill is not stable. Atoms are usually not stable when isolated. They usually combine to form more stable arrangements of matter.

~~CHAPTER 6 Chemical Bonding~~

molar mass of $C_6H_6O = 94.11 \text{ g}$ 21. a. molar
mass of $SO_3 = 80.07 \text{ g}$ 49.2 mg SO_3 1 g 1000 mg

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1 mol 80.07 g = 6.14 10^{-4} mol SO₃ b. molar mass of PbO₂ = 239.2 g 7.44 x 10⁴ kg PbO 2 1000 g 1 kg 1 mol 239.2 g = 3.11 10⁵ mol PbO 2 c. molar mass of CHCl₃ = 119.37 g 59.1 g CHCl₃ 1 mol 119.37 g = 0.495 mol CHCl₃ d. molar mass of C₂H₃Cl₃ = 133.39 g 3.27 ...

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Bonding SECTION 1 SHORT ANSWER Answer the

following questions in the space provided. 1.

a A chemical bond between atoms results from

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the attraction between the valence electrons and of different atoms.

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is tetrahedral questionprimary alkyl halide
answerOne carbon attached ... Test Answers on
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